

## UTILIZATION OF BICARBONATE BY APPLE FRUIT PHOSPHOENOLPYRUVATE CARBOXYLASE

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**Key Word Index**—*Malus domestica*; Rosaceae; apple; fruit; photosynthesis; phosphoenolpyruvate carboxylase.

**Abstract**—Phosphoenolpyruvate carboxylase (PEPC), was partially purified from apple fruit cv Golden Delicious. Kinetic values for PEP and  $\text{HCO}_3^-$  suggest a capacity for efficient carbon dioxide refixation. PEPC activity was maximal between 5–10 mM carbonate ( $\text{HCO}_3^-$ ) and inhibition was observed above 10 mM  $\text{HCO}_3^-$ . In conditions of PEP-saturation,  $\text{HCO}_3^-$  inhibition of apple fruit PEPC activity appeared non-competitive with respect to PEP and was partially reversible.

### INTRODUCTION

In a preliminary study [1] of the photosynthetic character of apple (*Malus domestica* Borkh) fruit, phosphoenolpyruvate carboxylase (PEPC) (E.C. 4.1.1.31) was identified in the fruit tissue. Phan [2] suggested  $\text{C}_4$  photosynthesis in apple fruit was based on the presence and activity of  $\text{C}_4$  organic acids and enzymes, interpretations of chloroplast morphology and the existence of a carbon dioxide concentrating mechanism. Apple fruit PEPC and chloroplasts have been found to differ in type from those found in  $\text{C}_4$  photosynthetic tissue [1, 3]. In an attempt to clarify the photosynthetic position of apple fruit we studied PEPC particularly in relation to kinetic constants for  $\text{HCO}_3^-$ , relating these to the known physiological concentrations. In fruit, but not in leaf tissue, internal carbon dioxide concentration may reach 1–5% [4–6]; physiological  $\text{HCO}_3^-$  at these concentrations may become relevant for inhibition of PEPC activity.

### RESULTS AND DISCUSSION

The kinetics of apple fruit PEPC with respect to  $K_m$  (PEP,  $\text{HCO}_3^-$ ) and  $K_i$  ( $\text{HCO}_3^-$ ) are shown in Table 1. Apple fruit PEPC appears to be an efficient carboxylating enzyme with low  $K_m$  (PEP) and a large capability for refixing respired carbon dioxide. The kinetic experiments were done with the aid of the coupled malate dehydrogenase (E.C. 1.1.1.31) (MDH) reaction (see Experimental), which was found to be insensitive to the  $\text{HCO}_3^-$  concentration used.

The kinetic constants (Table 1) were measured in conditions when the enzyme was saturated for PEP, i.e.  $20 \times K_m$ . However, at subsaturating PEP concentrations, apple fruit PEPC showed substrate competition between PEP and  $\text{HCO}_3^-$  at  $K_m$  (PEP) concentration and a non-competitive behaviour above a concentration of 0.2 mM PEP (Fig. 1). This effect is similar to that previously reported with PEPC from potato tubers [7].

Maximum PEPC activity of apple fruit was obtained at 5–10 mM  $\text{HCO}_3^-$  (Fig. 2) and inhibition of activity started above 10 mM, a concentration which corresponds to ca 5%  $\text{CO}_2$  at pH 7.8 which is the optimum pH for this tissue [8]. In apple fruit, internal carbon dioxide rises to 1–5% at maturity [4–6] which is below the starting point for inhibition of PEPC. Consequently under physiological conditions, inhibition by  $\text{HCO}_3^-$  would occur only rarely even in tissue with excess carbon dioxide. The comparative  $\text{HCO}_3^-$  inhibition data essentially confirm observations with single  $\text{HCO}_3^-$  concentrations, which were reported to be inhibitory, such as 3–5% carbon dioxide [9] and 30 mM  $\text{HCO}_3^-$  [10] with CAM tissue.

Partially purified apple fruit PEPC was assayed prior to and after addition of  $\text{HCO}_3^-$  at a concentration which caused 50% inhibition of activity. Excess  $\text{HCO}_3^-$  was then removed from the extracts by gel filtration. Relative to the initial activity under optimum conditions (5 mM  $\text{HCO}_3^-$ ), large  $\text{HCO}_3^-$  concentrations (250 mM) depressed the PEPC activity to 55% and this was restored to 88% by removal of the  $\text{HCO}_3^-$  (activity expressed/mg protein), demonstrating the reversibility of the inhibition.

Table 1. Kinetics of apple fruit PEPC

Tissue	$K_m(\text{PEP})$ (mM)	$K_m(\text{HCO}_3^-)$ (mM)	$K_i(\text{HCO}_3^-)$ (mM)	$V_{\max}$	
				(nkat /g fr. wt)	(nkat /mg protein)
Fruit	0.09	0.20	106	1.8	4.2
Seeds	0.09	0.20	115	5.4	10.8

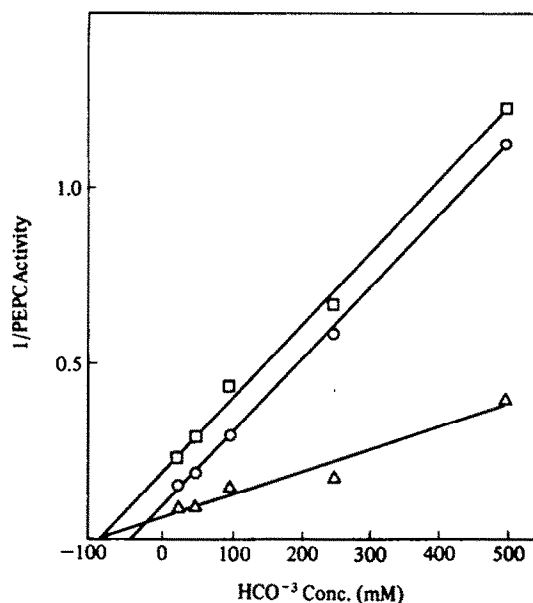


Fig. 1. PEPC kinetics. Interaction of the inhibition of apple fruit PEPC activity by large bicarbonate concentrations at various PEP substrate levels. PEP  $1 \times K_m$  ( $\square$ ),  $2 \times K_m$  ( $\circ$ ),  $20 \times K_m$  ( $\triangle$ ).

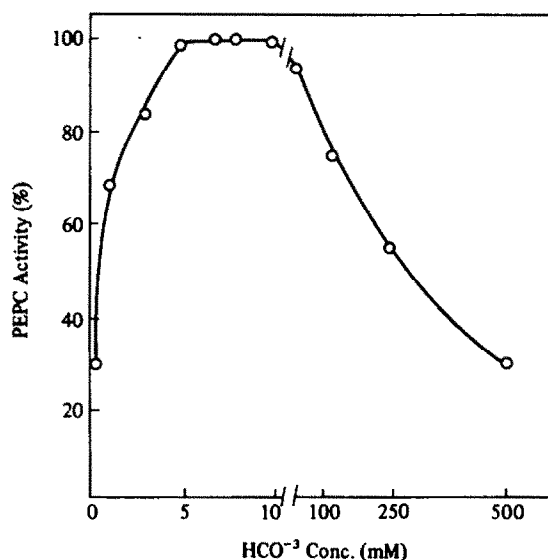


Fig. 2. The response of PEPC activity to bicarbonate concentration.

#### EXPERIMENTAL

**Plant material.** Apple fruit cv Golden Delicious were field grown at Long Ashton on 8 year old EMLA 106 rootstock.

**Preparation of extracts and enzyme assays.** PEPC was partially purified as previously reported [1]. Activity was measured spectrophotometrically at 340 nm by coupling the reaction to the

oxidation of NADH from MDH. The assay medium contained 50 mM Tris-HCl pH 7.8, 5 mM  $MgCl_2$ , 0.25 mM EDTA, 2 mM DTT, 10 units of MDH, 0.1 mM NADH and variable amounts of  $HCO_3^-$  (5 mM in the standard assay). The reaction was started by addition of PEP.

**Evaluation of kinetic constants.** Initial rate measurements, using the above assay, were recorded using a double beam spectrophotometer linked to a microcomputer.  $K_m$  and  $V_{max}$  values were calculated using a computer programme (Hucklesby, D. P. unpublished) based on a least squares analysis of data after Hanes transformation of  $s/v$  against  $s$  [11]. Estimates were also made from the same data and computer programme by direct linear plot [12] selecting the median values for  $K_m$  and  $V_{max}$  [12]. The two methods gave good agreement.  $K_i$  was measured using a graphical method [11] as  $1/v$  against  $i$ . Values given are derived from at least two repetitions of two extracts with at least six substrate concns. With PEPC,  $HCO_3^-$  serves as a substrate at small and an inhibitor at large concns. For each expt, the  $HCO_3^-$  concn giving max PEPC activity was evaluated to define the range of  $HCO_3^-$  concns subsequently used in the kinetic assays. All chemicals were made up in Tris buffer adjusted to pH 7.8 at room temp. at which the calculated endogenous  $HCO_3^-$  concn, following Hasselbach-Henderson equations [8], was 0.3 mM  $HCO_3^-$ , this value was integrated into all data presented in this paper.

**MDH activity** was measured spectrophotometrically at 340 nm in the direction of oxaloacetic acid reduction using a standard procedure [13]. **Protein** was measured using a protein assay reagent (BioRad) according to the manufacturer's instructions with ovalbumin as a standard.

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